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Key indicators

Single-crystal X-ray study T = 150 KMean σ (C–C) = 0.003 Å R factor = 0.046 wR factor = 0.042 Data-to-parameter ratio = 20.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. The title compound, $[Li_2(C_{13}H_{22}N_2Si)_2(C_4H_{10}O)_2]$, possesses a dinuclear structure featuring four-coordinate Li atoms, each of which shows a further, weak, contact to the central C atom of the NCN linkage of the amidinate ligand. The molecules possess crystallographically imposed inversion symmetry.

Bis(*µ*-*N*-*n*-propyl-*N*-trimethylsilylbenz-

amidinato)bis[(diethyl ether-O)lithium]

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Comment

In recent years, there has been much interest in N-silylated benzamidinates (Edelmann, 1994). New types of amidinate ligands with pendant amine or pyridine functionalities have recently been reported (Brandsma et al., 1998; Doyle et al., 2000; Kincaid et al., 1999). We have been interested in developing the chemistry of pendant-arm-functionalized amidinates as supporting ligands for group 4 imido chemistry (Boyd et al., 2002). As part of these studies, we were interested in using an N-propyl-substituted benzamidinate ligand, namely Me₃SiNC(Ph)NCH₂CH₂CH₃, for comparison with the pendant alkylamine-substituted amidinates Me₃SiNC(Ph)-NCH₂CH₂NMe₂ and Me₃SiNC(Ph)NCH₂CH₂CH₂NMe₂ (Brandsma et al., 1998; Doyle et al., 2000). We describe here the structure of the diethyl ether adduct of the lithium salt of this ligand, (I).



Molecules of (I) adopt a dinuclear structure in the solid state, possessing crystallographically imposed inversion symmetry. Each Li atom, and each propyl-substituted benzamidinate N atom, is four-coordinate, with Li deviating the most from an ideal tetrahedral geometry due to the restricted bite angle of the amidinate group. The central motif of the structure is a four-membered ring of alternating Li and N atoms, which is planar by symmetry. To either end of this ring is fused another, slightly puckered, four-membered ring. This puckering sees atoms C1 and Li lying out of the Li1/N1/C1/N2 least-squares plane of the ring by ca 0.19 and 0.07 Å, respectively. This creates a close Li1...C1 contact of 2.334 (3) Å, which is possibly electrostatic in nature. This distance is at the lower end of Li. · · CN2 distances observed for other lithiated amidinates, which are in the range ca 2.30-2.42 Å (Fletcher et al., 1996). The Li-N distances and N-Li-N angles in (I)

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metal-organic papers

span the typical ranges reported for lithiated amides (Fletcher *et al.*, 1996; Allen & Kennard, 1993).

The structural motif in (I) is closely related to that of the Li salt of the pendant *N*-propylamine-substituted ligand, namely $Li_2{Me_3SiNC(Ph)NCH_2CH_2CH_2NMe_2}_2$, synthesized by Lappert and co-workers (Doyle *et al.*, 2000). In this case, there is no Et₂O coordinated to Li and the remaining site is occupied by the chelating pendant amine N-donor.

Experimental

The title compound was prepared according to previously described procedures (Boyd *et al.*, 2002). Crystallization of the crude product from a mixture of hexane and Et_2O afforded the title compound as air-sensitive colourless blocks.

Z = 1

 $\theta = 5-27^\circ$

T = 150 K

 $\begin{aligned} R_{\rm int} &= 0.02\\ \theta_{\rm max} &= 27.5^\circ \end{aligned}$

 $h = -10 \rightarrow 10$

 $k = -13 \rightarrow 13$

 $l = -14 \rightarrow 15$

 $D_x = 1.078 \text{ Mg m}^{-3}$

Cell parameters from 3991

Mo $K\alpha$ radiation

reflections

 $\mu = 0.12 \text{ mm}^{-1}$

Block, colourless

 $0.32 \times 0.28 \times 0.28 \text{ mm}$

4372 independent reflections

2907 reflections with $I > 2\sigma(I)$

Crystal data

$$\begin{split} & \left[\text{Li}_2(\text{C}_{13}\text{H}_{22}\text{N}_2\text{Si})_2(\text{C}_4\text{H}_{10}\text{O})_2 \right] \\ & M_r = 628.94 \\ & \text{Triclinic, } P\overline{1} \\ & a = 8.4111 \ (17) \text{ \AA} \\ & b = 10.486 \ (2) \text{ \AA} \\ & c = 11.774 \ (2) \text{ \AA} \\ & \alpha = 106.59 \ (3)^{\circ} \\ & \beta = 96.05 \ (3)^{\circ} \\ & \gamma = 99.38 \ (3)^{\circ} \\ & \gamma = 99.38 \ (3)^{\circ} \\ & V = 969.2 \ (4) \text{ \AA}^3 \end{split}$$

Data collection

Enraf–Nonius KappaCCD diffractometer φ and ω scans Absorption correction: multi-scan (*DENZO*; Otwinowski & Minor, 1997) $T_{\min} = 0.97, T_{\max} = 0.97$ 7923 measured reflections

Refinement

Refinement on FH-atom parameters constrainedR = 0.046Weighting scheme: see textwR = 0.042 $(\Delta / \sigma)_{max} < 0.001$ S = 1.05 $\Delta \rho_{max} = 0.39$ e Å⁻³4117 reflections $\Delta \rho_{min} = -0.47$ e Å⁻³199 parameters $\Delta \rho_{min} = -0.47$ e Å⁻³

Table 1

Selected geometric parameters (Å, °).

Li1-N12.023 (3)N1-Si11.70Li1-N22.145 (3)N1-C11.33Li1-N2^i2.073 (3)N2-C11.33Li1-O11.965 (3)N2-C81.47Li1-C12.334 (3)C1-C21.51N1-Li1-N266.61 (11)Li1-N2-C1127.N1-Li1-N2121.29 (16)Li1-N2-C8141.N2-Li1-N2^i107.22 (14)Li1^i-N2-C8108.N1-Li1-O1123.08 (16)C1-N2-C8119.N2-Li1-O1111.49 (16)Li1-O1-C15134.N2'-Li1-O1113.47 (15)Li1-O1-C16109.	
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	3 (15)
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$\begin{array}{cccccccccccccccccccccccccccccccccccc$	4 (15)
$\begin{array}{ccccccc} N2-Li1-N2^{i} & 107.22 (14) & Li1^{i}-N2-C8 & 108. \\ N1-Li1-O1 & 123.08 (16) & C1-N2-C8 & 119. \\ N2-Li1-O1 & 111.49 (16) & Li1-O1-C15 & 134. \\ N2^{i}-Li1-O1 & 113.47 (15) & Li1-O1-C16 & 109. \\ N1-N1-N1 & S11 & 110.05 (12) & C15 & O1-C16 & 110. \\ \end{array}$	0(14)
$ \begin{array}{ccccccc} N1-Li1-O1 & 123.08 \ (16) & C1-N2-C8 & 119. \\ N2-Li1-O1 & 111.49 \ (16) & Li1-O1-C15 & 134. \\ N2^i-Li1-O1 & 113.47 \ (15) & Li1-O1-C16 & 109. \\ N1-N1-N1 & S11 & 140.55 \ (15) & C15 & O1 & O16 \ (15) & 110.55 \ (15) & C15 & C16 \ (15) \$	4 (14)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	1 (15)
$N2^{i}-Li1-O1$ 113.47 (15) $Li1-O1-C16$ 109.	8 (15)
L'1 N1 S'1 140.05 (12) C15 O1 C16 140	7 (14)
$L_{11} - N_{1} - S_{11}$ 140.05 (12) $C_{12} - O_{1} - C_{16}$ 112.	3 (15)
Li1-N1-C1 85.59 (14) N1-C1-N2 118.	1 (16)
Si1-N1-C1 131.63 (13) N1-C1-C2 119.	8 (15)
$Li1-N2-Li1^{i}$ 72.78 (14) $N2-C1-C2$ 121.	5 (15)
Li1-N2-C1 80.64 (13)	

Symmetry code: (i) 1 - x, 1 - y, 2 - z.

The weighting scheme used in the refinement was a Prince modified Chebychev polynomial (Watkin, 1994), in which



Figure 1

View of the molecular structure of (I). Displacement parameters are drawn at the 20% probability level and H atoms have been omitted for clarity. Atoms carrying the suffix A are related to their counterparts by the symmetry code (1 - x, 1 - y, 2 - z).

 $W = [\text{weight polynomial}][1 - (\Delta F/6\sigma F)^2]^2$, with coefficients -0.679, -2.35, -1.48, -0.984 and -0.429. All H atoms were placed geometrically and refined with a riding model.

Data collection: *COLLECT* (Nonius, 2000); cell refinement: *DENZO* (Otwinowski & Minor, 1997); data reduction: *DENZO*; program(s) used to solve structure: *SIR*97 (Altomare *et al.*, 1999); program(s) used to refine structure: *CRYSTALS* (Watkin *et al.*, 2001); molecular graphics: *CAMERON* (Watkin *et al.*, 2001); software used to prepare material for publication: *CRYSTALS*.

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References

- Allen, F. H. & Kennard, O. (1993). Chem. Des. Autom. News, 8, 1, 31-37.
- Altomare, A., Burla, M. C., Camalli, M., Cascarano, G., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Polidori, G. & Spagna, R. (1999). J. Appl. Cryst. 32, 115–119.
- Boyd, C. L., Guiducci, A. E., Dubberley, S. R., Tyrrell, B. R. & Mountford, P. (2002). J. Chem. Soc. Dalton Trans. In the press.
- Brandsma, M. J. R., Brussee, E. A. C., Meetsma, A., Hessen, B. & Teuben, J. H. (1998). *Eur. J. Inorg. Chem.* pp. 1867–1870.
- Doyle, D., Gun'ko, Y. K., Hitchcock, P. B. & Lappert, M. F. (2000). J. Chem. Soc. Dalton Trans. pp. 4093–4097.
- Edelmann, F. T. (1994). Coord. Chem. Rev. 137, 403-481.
- Fletcher, D. A., McMeeking, R. F. & Parkin, D. (1996). J. Chem. Inf. Comput. Sci. 36, 746–749.
- Kincaid, K., Gerlach, C. P., Giesbrecht, G. R., Hagadorn, J. R., Whitener, G. D., Shafir, A. & Arnold, J. (1999). Organometallics, 18, 5360–5366.
- Nonius (2000). COLLECT. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). Methods in Enzymology, Vol. 276, Macromolecular Crystallography, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.
- Watkin, D. J. (1994). Acta Cryst. A50, 411-437.
- Watkin, D. J., Prout, C. K., Carruthers, J. R., Betteridge, P. W. & Cooper, R. I. (2001). CRYSTALS, Issue 11. Chemical Crystallography Laboratory, Oxford, England.
- Watkin, D. J., Prout, C. K. & Pearce, L. J. (1996). CAMERON. Chemical Crystallography Laboratory, University of Oxford, Oxford.